The ¹H NMR Spectral Detection and Kinetic Analysis of the Conformational Change of 4,4,5,5- and 6,6,7,7-Tetramethoxycycloheptadienones

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The variable temperature ¹H NMR measurements of 4,4,5,5- and 6,6,7,7-tetramethoxycycloheptadienones disclosed an existence of the bond-twisting conformers. The complete line shape analysis by the simulation method gave the activation parameters of the interconversion barrier.

When we prepared o- and p-tropoquinone bisacetals by means of an anodic oxidation of dimethoxytropones, 1,2) we noticed that their 1 H and 13 C NMR spectra showed a considerable broadening of the MeO signals of bisacetals derived from 2,3- and 4,5-dimethoxytropones even at 20 °C due to a dynamic process. Recently, Weissensteiner et al. made a conformational analysis of 10,11-dihydrodibenzo[a,d]cyclohepten-5-one by means of X-ray diffraction and NMR spectroscopic studies. 3 Herein, we describe the kinetic aspects of the interconversion of our monocyclic tetramethoxycycloheptadienones.

The ¹H NMR of 4,4,5,5-tetramethoxycyclohepta-2,6-diencnes (1) at 25 °C showed two broad MeO signals, while that of 6,6,7,7-tetramethoxycyclohepta-2,4-dienone (2) showed a sharp signal for two MeO and a broad signal for two MeO. Variable temperature measurement of the ¹H NMR of 1 (between -60 °C and 60 °C) and complete line shape analysis by the simulation method⁴⁾ provided kinetic parameters listed in Table 1.

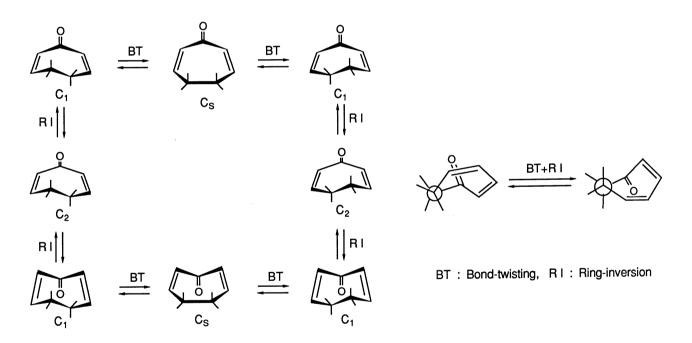
There are two molecular motions in the system of $1;^3$ i.e., the flipping of the carbonyl group (ring inversion) of the cycloheptadienone via a C_2 symmetrical transition state and the bond-twisting at the sp³-carbons to exchange between two half-boat forms (bond-twisting) via a C_s symmetrical transition state. 4,5-Ethylenedioxy-4,5-dimethoxycyclohepta-2,6-dienone ($\mathbf{3}$)⁵ showed no broadening of

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the MeO signal even at -60 °C, confirming the rapid ring inversion. Since two MeO signals of 1 appeared separately at -60 °C, the cycloheptadienone is operating a rapid ring inversion but no twisting of the C-C bond. On the other hand, 2 behaved differently; the 1 H NMR of 2 at -30 °C showed four MeO signals, indicating a frozen molecular motion. An appearance of four MeO signals in 2 can be explained in terms of fixed twisted-boat conformer. The ring inversion and the bond twisting operated simultaneously in 2. All $^{\Delta}$ H values of 1, ca. 56 kJ mol $^{-1}$, were larger than that of 2, and no substituent effect of the dialkyl groups on the C-2 and C-7 was observed.

		1		
	∆H [≠] /kJ mol ⁻¹	$\Delta S^{\neq}/J \text{ mol}^{-1} K^{-1}$	∆G [≠] /kJ mo1 ⁻¹ at 298 K	r
1a	56.5±1.3	-20.5±4.2	62.7±2.5	0.9996
1 b	59.0±2.9	-18.8±9.2	64.8±5.9	0.9985
1c	56.9±1.3	-25.5±3.3	64.4±2.1	0.9998
2	42.6±2.5	-45.6±10.0	56.4±5.9	0.9956

Table 1. Activation parameters for the interconversion



A more rapid conformational change of 2 is explained in that the ring inversion can make the bond twisting easy because two exchanging parts are directly connected. The larger value of ΔS^{\neq} is probably due to that the exchanging part in 2 is wider than that in 1.

References

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